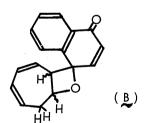
## PHOTOCHEMICAL REACTION OF CYCLOHEPTATRIENE WITH SOME AROMATIC QUINONES

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The photochemical addition reaction of cycloheptatriene with <u>p</u>-benzoquinone and 1,4-naphthoquinone yielded the spiro-ethers having 7-oxa-bicyclo[4,2,1] nonadiene moiety by the characteristic  $(6+2)\pi$  addition process. The latter quinone further produced the carbocyclic  $(2+2)\pi$  - and  $(6+2)\pi$  cycloadducts. Either  $(4+2)\pi$  adducts or dioxetanes were undetectable in both cases. The photoadducts were stable under their formation conditions, showing no tendency to cause the isomerization.

As a part of our series of studies of the photocycloaddition reactions of conjugated olefins with various carbonyl functions, 1) we have extended the investigation to that of cycloheptatriene (1) with two quinones, p-benzoquinone (2) and 1,4-naphthoquinone (3), in order to know the pericyclic mode of the reaction in general, and to check the photochemical hydrogen abstraction to give ene-products or bis-cycloheptatrienyl (A) in connections with the ground state properties of quinones. On the photocycloaddition reaction of 1 with 3, in fact Gandhi et al. 2) have already reported the results which described a formation of a 1:1-adduct,



as bright-red crystals, whose structure has been claimed to be a (2+2)  $\pi$  adduct (B) derived from the carbonyl chromophore, on the basis of
100 MHz NMDR spectral evidences. The purpose of this communication
(B) is to show our own results on the photocycloaddition of 1 and 3 in
addition to that of 1 and 2.3)

When a benzene solution of 1 and  $2^4$ ) was irradiated by means of a 450 W high-pressure mercury lamp through a Pyrex glass filter, an oily 1:1-cycloadduct (m/e:200)(4) was obtained in 56 % yield. No other product was detectable despite an intensive survey for the every chromatographic fraction. The structure of 4 was determined as follows:The NMR [6:2.75(3H, m), 4.70(1H, t, J=6.5 Hz), 5.90(6H, overlapped m),

6.71(1H, dd, J=9.5, 3.0 Hz), and 7.01(1H, dd, J=9.5, 3.0 Hz)] spectrum revealed a presence of the dienone moiety as like as that of the reported cyclooctatetraene-2 adduct ( $\Sigma$ ), however, an incidental overlapping of the aliphatic proton signals made impossible to discern the structure to determine the pericyclic mode of the addition, (6+2) $\pi$ -, (4+2) $\pi$ - or (2+2) $\pi$  process. The NMDR [ $\delta$ :3.21(1H, d, J=12.8 Hz), 3.64(1H, dt, J=12.8, 6.5 Hz), 4.00(1H, t, J=6.5 Hz), 5.48(1H, t, J=6.5 Hz), 6.40(4H, m), 8.22(1H, dd, J=9.5, 3.0 Hz), 8.58(1H, dd, J=9.5, 3.0 Hz), 10.64(1H, d, J=9.5 Hz), and 10.76(1H, d, J=9.5 Hz)] spectrum in the presence of a shift reagent [ $\frac{4}{3}$ (31.4 mg)/Eu(DPM) $_3$ (60.7 mg)] revealed that among the four aliphatic protons, two allylic methine protons did not spin-couple in each other, and the observed magnitude of the gem-coupling constant for the non-allylic methine protons (12.8 Hz) should be appropriate for that of a tetrahydrofuran. Therefore,  $\frac{4}{3}$  must be a (6+2) $\pi$  cycloadduct. The observed UV spectrum ( $\lambda_{\text{max}}^{\text{MeOH}}$ :228 nm (E:15500), 264.5(4400)) $^6$ ) was also in accord with this view. When an ethanolic solution of  $\frac{4}{3}$  was treated with small amounts of hydrochloric acid at room temperature, a phenolic compound ( $\frac{5}{3}$ , mp 85-87°C) and its ethyl ether ( $\frac{6}{3}$ , a colorless oil) were obtained in 23 % and 56 % yields respectively. Again, the NMR spectra of  $\frac{5}{3}$ [ $\delta$ :1.98(1H, ddd, J=14.0, 2.0, 1.0 Hz), 2.56(1H, ddd, J=14.0, 6.5, 4.0 Hz), 3.68(1H, dd, J=8.5, 6.5 Hz), 4.9(1H, OH), 4.94(1H, m),

5.90(3H, m), and 6.3-6.8(4H, m)] and  $\underline{6}$  [  $\underline{6}$ :1.33(3H, t), 1.96(1H, ddd, J=14.0, 2.5, 1.2 Hz), 2.58(1H, ddd, J=14.0, 6.5, 4.5 Hz), 3.65(1H, br. t, J=6.5 Hz), 3.93(2H, q), 4.93(1H, m), 5.7-6.1(3H, m), and 6.2-6.9(4H, m)] clearly excluded the (2+2) $\pi$  structure.

On the other hand, the reaction of 1 and 3 under the similar conditions with the above showed rather complicated features; a formation of a stereoisomeric mixture of  $(6+2)\pi$  ethers (7, a colorless oil (35%) and 8, a colorless oil (2.5%)), a  $(2+2)\pi$  cycloadduct  $(9, \text{ pale yellow needles, mp } 103-104^{\circ}\text{C}(10\%))$  and a  $(6+2)\pi$  cycloadduct  $(10, \text{ pale yellow needles, mp } 87-88^{\circ}\text{C}(13\%))$  together with a dehydro-derivative of  $(10, \text{ pale yellow needles, mp } 158-160^{\circ}\text{C}(4\%))$  was recognized. These products were identified by the spectroscopic and chemical methods of analyses.

The NMR spectra of \$\int\_{1}^{2} \{ 6:2.68(1H, d, J=12.5 Hz), 2.73(1H, t, J=7.0 Hz), 3.00(1H, dt, J=12.5, 7.0 Hz), 4.76(1H, t, J=7.0 Hz), 5.10(1H, dd, J=11.0, 7.0 Hz), 5.6-6.2(3H, m), 6.05(1H, d, J=10.5 Hz), 7.18(1H, d, J=10.5 Hz), and 7.24-8.00(4H, m)] and of \$\{ 6:2.40(1H, d, J=12.5 Hz), 2.5-3.1(2H, m), 4.87(1H, br. t, J=7.0 Hz), 5.7-6.4(4H, m), 6.18(1H, d, J=10.5 Hz), 7.30(1H, d, J=10.5 Hz), and 7.2-8.1(4H, m)] were closely resembled in each other, and comparisons of chemical shift of the corresponding pairs of hydrogens, i. e. one of the methylene protons of \$\{ 8(2.40)\$ appeared in higher field than that of \$\int\_{1}^{2}(2.68)\$, and one of the four vinyl protons of \$\int\_{2}(5.10)\$ caused a marked up-field shift, while the four protons of \$\{ 8(5.7-6.4)\$ behaved as usual, the stereostructures of \$\int\_{2}\$ and \$\{ 8\}\$ were deduced as indicated in the chart. It is interesting that the sterically more clowded isomer, \$\int\_{2}\$, was predominant over the other. \$\int\_{2}\$ also gave an aromatized ethyl ether (12) by treatment with hydrochloric acid in ethanol at room temperature, while \$\{ 8\}\$ needed to reflux for 2 hr to give the same product. The NMR [\$\{ 6:1.47(3H, t), 2.04(1H, ddd, J=13.8, 2.5, 1.0 Hz), 2.68(1H, ddd, J=13.8, 5.5, 4.0 Hz), 3.68(1H, dddd, J=8.5, 5.5, 2.0, 1.0 Hz), 4.08(2H, q), 5.14(1H, m), 5.7-6.1(3H, m), 6.2-6.5(1H, m), 6.40(1H, s), 7.3-7.6(2H, m), and 8.0-8.3(2H, m)] spectrum of 12( colorless plates, mp 129-130°C) again confirmed the pericyclic mode of the addition process; especially, an observed change of the magnitude of the gem-coupling constant for the methylene protons is in a range of the expectation.

9 [ 6:2.49(1H, ddd, J=13.0, 6.5, 4.0 Hz), 2.74(1H, ddd, J=13.0, 9.0, 4.0 Hz), 3.0-3.6(3H, m), 3.80(1H, dd, J=9.0, 7.5 Hz), 5.95(4H, m), and 7.6-8.2(4H, m)] showed rather poorly resolved NMR spectrum, but the methylene protons were proven to be at an allylic carbon by spin-decoupling experiments, thus, 9 can not be a (6+2)% cycloadduct. A catalytic reduction of 9 with platinum oxide gave an octahydro-derivative (13, colorless needles, mp 248-250°C) which was identical with the tetrahydro-derivative of the one (D) of the photoadducts of 3 with cycloheptene. Therefore, 9 is a (2+2)% cycloadduct.

On the other hand, 10 [ 6:1.87(1H, br. dt, J=13.0, 2.0 Hz), 2.04(1H, dtm, J=13.0, 6.0 Hz), 3.25(2H, br. dd, J=7.5, 6.0 Hz), 3.74(2H, br. s), 5.84(2H, dm, J=12.0 Hz), 6.0-6.2(2H, dm, J=12.0 Hz), 7.6-7.9(2H, m), and 8.0-8.2(2H, m)] was deduced to be a symmetrical  $exo-(6+2)\pi$  cycloadduct on the basis of the coupling patterns of protons at the ring juncture. 11, prepared also by dehydrogenation of 10 with 2,3-dichloro-5,6-dicyanobenzoquinone, exhibited also a symmetrical NMR spectrum[6:1.69(1H, d, J=12.0 Hz), 2.36(1H, dtt, J=12.0, 7.0, 1.5 Hz), 3.98(2H, t, J=7.0 Hz), 5.9(2H, m), 6.2(2H, m), 7.6(2H, m), and 8.2(2H, m)].

It is remarkable that the formation of cyclic ethers is not stereospecific but perispecific, while, the formation of carbocyclic products is stereospecific but not perispecific. The oxetane, and the ene-products or those related product, A, were undetectable. No  $(4+2)\pi$  cycloaddition process has occurred. With some contrasting features, the reaction of 1 with methyl acetopyruvate has afforded the stereospecific  $(4+2)\pi$  and  $(6+2)\pi$  cycloadducts and the ene-product.

When this cycloaddition was performed by means of a 500 W tungsten lamp, the composition of the photo-products showed no difference with that by a mercury lamp, and the product ratio was constant during the irradiation.

Although our reaction conditions seem to cover those of Gandhi,  $^{2)}$  we have failed to identify the crystalline C.

## References and Notes

- 1) H. Takeshita, A. Mori, and Y. Toyonaga, Bull. Chem. Soc. Japan, 48, 307 (1975).
- 2) R. P. Gandhi, S. N. Dhawan, and S. M. Mukherji, Ind. J. Chem., 9, 283 (1971).
- 3) An attempted reaction of 1 with chloranil under the similar conditions gave no adduct.
- 4) All the experiments in this paper were undertaken as one mole solution of quinones with two- to five-fold excess of lat 10-15°C under nitrogen atmosphere. The compounds described here gave satisfactory elemental analyses.

  The NMR spectra were measured in CDCl<sub>3</sub> solutions.
- 5) D. Bryce-Smith, A. Gilbert, and M. G. Johnson, J. Chem. Soc., C, 383 (1967).
- 6) The structure of <u>C</u> has recently been revised to the (6+2) t structure. <u>Cf.</u> R. M. Wilson, E. J. Gardner, R. C. Elder, R. H. Squire, and L.R. Florian, <u>J. Amer. Chem. Soc.</u>, 96, 2955 (1974). They reported λ max of <u>C</u>: 199 nm(£:11400), 223(16200), 260(5400).
- 7) Photocycloaddition of 3 with cycloheptene gave the carbocyclic (2+2)  $\pi$  adduct, D, and an isomeric mixture of dioxetanes.
- 8) H. Takeshita, A. Mori, and S. Itô, Bull. Chem. Soc. Japan, 47, 1767 (1974).
- 9) Generally, the irradiations by tungsten lamp gave better results due to diminishment of polymerization; at one instance, 1 and 2 gave 4 in 68 % yield.

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